

Convolutamines A-E, Novel  $\beta$ -Phenylethylamine Alkaloids from Marine Bryozoan *Amathia convoluta*Hui-ping ZHANG, Yoshiaki KAMANO,\* Haruhisa KIZU,<sup>†</sup>Hideji ITOKAWA,<sup>††</sup> George R. PETTIT,<sup>†††</sup> and Cherry L. HERALD<sup>†††</sup>

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Convolutamines A-E, novel cytotoxic alkaloids, have been isolated from Floridian bryozoan *Amathia convoluta* and the structures have been elucidated on the basis of extensive spectroscopic data.

Chemical studies of marine bryozoans have provided a variety of interesting bioactive secondary metabolites including alkaloids from *Amathia wilsoni*<sup>1)</sup> and from *Flustra foliacea*<sup>2)</sup> as well as antineoplastic macrolides bryostatins from *Bugula neritina*<sup>3)</sup> and from *Amathia convoluta*.<sup>4)</sup> During our search for new biologically active substances from marine organisms, we have examined an extract of the Floridian bryozoan *Amathia convoluta* collected off the Northeastern Gulf of Mexico. The EtOAc-soluble material from the extract was subjected to repeated chromatographies on silica gel, ODS, and Sephadex LH-20 to afford five novel  $\beta$ -phenylethylamine alkaloids convolutamines A (**1**), B (**2**), C (**3**), D (**4**), and E (**5**), as colorless oils, in the yields of  $5.5 \times 10^{-4}\%$ ,  $2.5 \times 10^{-5}\%$ ,  $1.7 \times 10^{-6}\%$ ,  $1.4 \times 10^{-5}\%$ , and  $4.0 \times 10^{-7}\%$ , respectively. We describe herein the structural elucidation of them.

The EIMS spectrum of convolutamine A (**1**)<sup>5)</sup> showed ions  $(M-CH_3)^+$  at  $m/z$  442, 444, 446, and 448 in the ratio of 1:3:3:1, suggesting the presence of three bromine atoms. The molecular formula of **1** was determined as  $C_{13}H_{18}O_2NBr_3$  by HREIMS [ $m/z$  456.8929 ( $M$ )<sup>+</sup> for  $C_{13}H_{18}O_2N^{79}Br_3$ ;  $\Delta$  +4.1mmu], indicating four degrees of unsaturation. The <sup>1</sup>H<sup>5)</sup> and <sup>13</sup>C (Table 1) NMR spectra in combination with <sup>1</sup>H-<sup>1</sup>H COSY, HMQC, HMBC, INADEQUATE experiments showed that **1** consisted of an aromatic ring and an aliphatic chain, as depicted in Fig. 1. The <sup>1</sup>H and <sup>13</sup>C NMR signals for the aromatic portion (C-6-C-11) in **1** suggested the presence of a 1, 2, 3, 4, 6-pentasubstituted benzene ring, which was verified by the HMBC cross-peaks for H-10/C-6, H-10/C-8, H-10/C-9, and H-10/C-11. This result was further confirmed on the basis of <sup>1</sup>H nondecoupling <sup>13</sup>C NMR data, which were found to be  $^3J_{CH} = 7.4$  Hz for H-10/C-8 and  $^2J_{CH} = 4.4$  Hz for both H-10/C-9 and H-10/C-11.<sup>6)</sup> Therefore the quaternary aromatic carbons at  $\delta_C$  121.6s, 115.9s, and 119.7s could be respectively assigned to C-7, C-9, and C-11 to the result that bromine atoms were substituted at C-7, C-9, and C-11. On the other hand, the signal at 119.7 ppm must be located at C-11 position, because the correlations for C-9/C-10 and C-10/C-11 were observed in the INADEQUATE spectrum. The HMBC cross-peak for H-12/C-8 indicated that the methoxyl group was attached to C-8. The partial structure of the aliphatic chain (C-1-C-5) in **1**

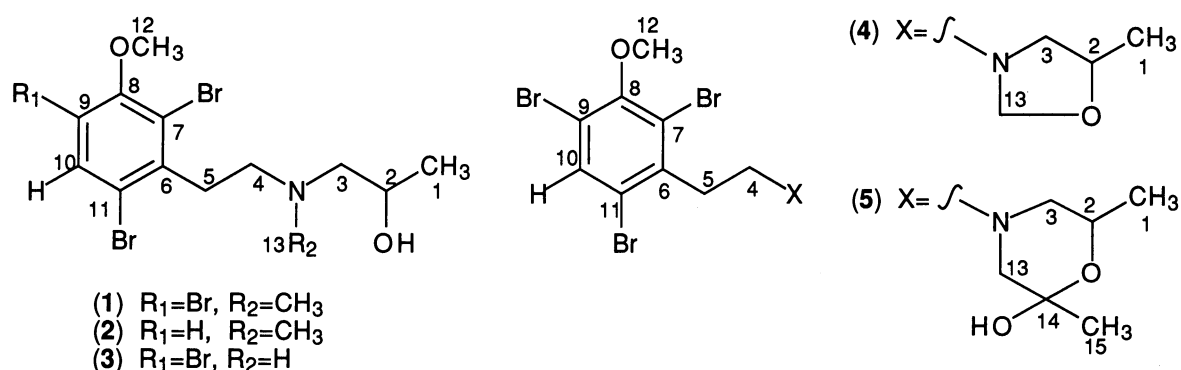


Fig. 1 The structures of convolutamines A (1), B (2), C (3), D (4), and E (5).

was deduced from the COSY correlations for H-1/H-2, H-2/H-3, and H-4/H-5 as well as the HMBC cross-peaks for H-13/C-3 and H-13/C-4. The tribromomethoxyphenyl unit (C-6-C-11) was shown to be connected to C-5 position by the HMBC cross-peaks for H-4/C-6, H-5/C-6, H-5/C-7, and H-5/C-11. An  $^1\text{H}$  NMR signal due to the hydroxyl group on C-2 was observed at  $\delta_{\text{H}}$  4.42d in  $\text{DMSO}-d_6$  and disappeared by addition of  $\text{D}_2\text{O}$ . This was also supported by the IR absorption band at  $3450\text{ cm}^{-1}$ . These results led to structure 1, 3-[N-methyl-2-(2,4,6-tribromo-3-methoxyphenyl)ethylamino]-2-propanol, for convolutamine A, as shown in Fig. 1.

Table 1.  $^{13}\text{C}$  NMR Data of Convolutamines A (1), B (2), C (3), D (4), and E (5) in  $\text{CDCl}_3$

position	1	2	3	4	5
1	19.9 q	19.6 q	20.4 q	20.0 q	18.9 q
2	63.1 d	63.0 d	65.5 d	71.7 d	65.3 d
3	64.7 t	64.7 t	56.3 t	59.5 t	59.1 t
4	55.0 t	55.1 t	47.1 t	52.3 t	55.1 t
5	34.7 t	34.4 t	37.9 t	37.3 t	34.5 t
6	139.7 s	139.8 s	139.5 s	139.3 s	139.5 s
7	121.6 s	115.1 s	121.8 s	121.8 s	121.8 s
8	154.0 s	155.6 s	154.0 s	154.0 s	154.0 s
9	115.9 s	110.9 d	116.1 s	116.1 s	116.1 s
10	135.4 d	131.8 d	135.4 d	135.4 d	135.4 d
11	119.7 s	110.9 s	119.9 s	119.8 s	119.8 s
12	60.5 q	56.5 q	60.5 q	60.5 q	60.5 q
13	41.9 q	41.9 q		86.3 t	61.5 t
14					93.8 s
15					25.2 q

$\delta$  in ppm.

The spectral data of convolutamine B (2)<sup>7)</sup> were similar to those of 1. The EIMS spectrum revealed that 2 possessed one less bromine atom than 1, since the molecular ions at  $m/z$  379, 381, 383 showed a triplet in a ratio of 1:2:1, which suggested the presence of two bromine atoms. The molecular formula,  $\text{C}_{13}\text{H}_{19}\text{O}_2\text{NBr}_2$ , was deduced by HREIMS [ $m/z$  378.9784, (M)<sup>+</sup> for  $\text{C}_{13}\text{H}_{19}\text{O}_2\text{N}^{79}\text{Br}_2$ ,  $\Delta +0.1\text{mmu}$ ]. The  $^1\text{H}$  NMR spectrum of 2 closely resembled that of 1 except the signals at  $\delta_{\text{H}}$  6.65d and  $\delta_{\text{H}}$  7.45d. The  $^{13}\text{C}$  NMR spectral data (Table 1) of 2 were different from those of 1 only in the signals due to C-7, C-9, C-10, and C-11. Analysis of 2D NMR

spectral data revealed that **2** possessed the structure as follows. The  $^1\text{H}$ - $^1\text{H}$  coupling constant ( $J = 8.8 \text{ Hz}$ ) due to  $\delta_{\text{H}} 6.65\text{d}$  and  $7.45\text{d}$  implied that H-9 connected to ortho position of H-10, which was further confirmed by the COSY correlation between H-9 and H-10. Also, the position of H-9 was established by the NOESY correlation between H-9 and H<sub>3</sub>-12 as well as the HMBC cross peaks for H-9/C-7, H-9/C-8 and H-9/C-11. The  $^1\text{H}$  NMR spectral data of **2** in  $\text{DMSO}-d_6$  also revealed the presence of an exchangeable proton due to OH-2 by addition of  $\text{D}_2\text{O}$ . Thus, the structure of convolutamine B was determined to be structure **2**, 3-[*N*-methyl-2-(2,6-dibromo-3-methoxyphenyl)ethylamino]-2-propanol.

Convolutamine C (**3**)<sup>8)</sup> was isolated from a more polar portion than any of two compounds described above. The  $^1\text{H}$  ( $\delta_{\text{H}} 2.43 \text{ s}$ ) and  $^{13}\text{C}$  ( $\delta_{\text{C}} 41.9 \text{ q}$ ) NMR signals (Table 1) due to *N*-methyl group which observed in both **1** and **2** were absent in **3**. By the analyses of NMR spectral data, the structure of convolutamine C (**3**) was assigned to be structure **3**, 3-[2-(2,4,6-tribromo-3-methoxyphenyl)ethylamino]-2-propanol, as shown in Fig. 1.

In the  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of convolutamine D (**4**), the signals at  $\delta_{\text{H}} 4.41\text{d}$  and  $4.44\text{d}$  (H-13)<sup>9)</sup> as well as the signal at  $\delta_{\text{C}} 86.3\text{t}$  (C-13) (Table 1) were observed to be different from those of **1** and **2**. The HMBC cross peaks for H-13/C-4 and H-13/C-3 as well as the COSY correlation H-1-H-3 suggested the presence of an oxazolidine ring. The  $^{13}\text{C}$  chemical shift ( $\delta 86.3 \text{ t}$ ) for the methylene signal due to C-13 agreed well with that a oxazolidine system.<sup>10)</sup> The EIMS spectrum showed the molecular ions ( $\text{M}^+$ ) at  $m/z$  454, 456, 458, 460 in a ratio of 1:3:3:1, suggesting the presence of three bromine atoms as **1**. Interpretation of the spectral data such as  $^1\text{H}$ - $^1\text{H}$  COSY, HMQC, HMBC led to the establishment of the structure of **4** as shown in Fig. 1, assigned as 3-[2-(2,4,6-tribromo-3-methoxyphenyl)ethyl]-5-methyloxazolidine. To our knowledge, compounds with oxazolidine are rare in nature.

The EIMS spectrum for convolutamine E (**5**)<sup>11)</sup> showed ions  $(\text{M}-\text{H}_2\text{O})^+$  at  $m/z$  481, 483, 485, 487 in a ratio of 1:3:3:1, suggesting the presence of three bromine atoms. The molecular formula,  $\text{C}_{15}\text{H}_{20}\text{O}_3\text{NBr}_3$ , was established by the HREIMS[( $\text{M}$ )<sup>+</sup>,  $m/z$  498.8994, for  $\text{C}_{15}\text{H}_{20}\text{O}_3\text{N}^{79}\text{Br}_3$ ]. With the aid of NMR techniques such as  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR (Table 1), HOMO decoupling and NOE difference spectral experiments, the structure of **5** was assigned as 4-[2-(2,4,6-tribromo-3-methoxyphenyl)ethyl]-2-hydroxy-2,6-dimethylmorpholine, as shown in Fig. 1.

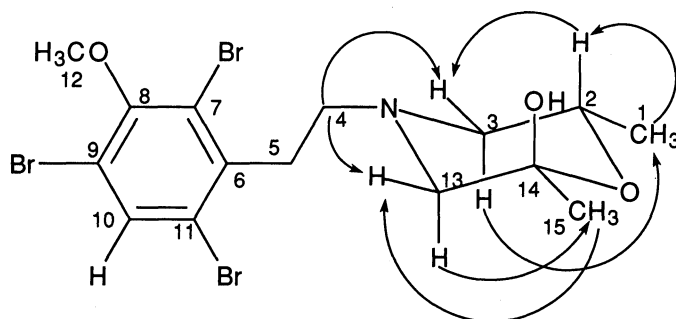


Fig. 2 NOEs of convolutamine E (**5**) in pyridine- $d_5$ .

The relative stereochemistry of convolutamine E (**5**) was illustrated in Fig. 2. Both convolutamines A(**1**) and B (**2**) were found to be a racemates, because the  $^1\text{H}$  NMR spectra of their 2-O-(*S*)- or (*R*)-MTPA

derivatives showed the signals due to both (2*S*)- and (2*R*)-form in a ratio of 1:1.<sup>12)</sup> The stereochemistry of convolutamines C and D remains unknown.

From other bryozoan *Amathia wilsoni*, a  $\beta$ -phenylethylamine has been isolated as a possible precursor of the brominated  $\beta$ -phenylethylamines, amathiamides, having the *N*-methylpyrrolidine groups.<sup>1)</sup> This fact indicated that the biogenesis of convolutamines A–E (1–5) from *Amathia convoluta* is related to that of Amathiamides.

Convolutamines A (1), B (2), and D (4) exhibited cell growth inhibitory activity (IC<sub>50</sub> 10.6, 4.8, and 8.6  $\mu$ g/ml, respectively) against the murine P388 lymphocytic leukemia.

#### References

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- 5) 1: IR (CHCl<sub>3</sub>)  $\nu_{\max}$  3450, 1460, 1420, 1350, 1275, 1050, 936 cm<sup>-1</sup>. UV (MeOH)  $\lambda_{\max}$  212.5 nm ( $\epsilon$  27, 900). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  1.15 (3H, d,  $J$  = 5.9 Hz, H-1), 2.34 (1H, dd,  $J$  = 12.1, 10.3 Hz, H-3a), 2.44 (1H, dd,  $J$  = 12.1, 2.9 Hz, H-3b), 2.44 (3H, s, H-13), 2.59 (1H, ddd,  $J$  = 12.5, 11.0, 5.5 Hz, H-4a), 2.71 (1H, ddd,  $J$  = 12.5, 11.0, 5.5 Hz, H-4b), 3.13 (1H, ddd,  $J$  = 12.5, 11.0, 5.5 Hz, H-5a), 3.21 (1H, ddd,  $J$  = 12.5, 11.0, 5.5 Hz, H-5b), 3.81 (1H, dqd,  $J$  = 10.3, 5.9, 2.9 Hz, H-2), 3.86 (3H, s, H-12), 7.74 (1H, s, H-10).
- 6) The C-H long range coupling constants of aromatic carbons:  $^3J_{\text{CH}} > ^2J_{\text{CH}}$ , e.g.  $^3J_{\text{CH}} = 7.4$  Hz and  $^2J_{\text{CH}} = 1.0$  Hz (benzene ring). R. M. Silverstein, G. C. Bassler, and T. C. Morrill, "Spectrometric Identification of Organic Compounds", Fourth Edition (1981), p. 272.
- 7) 2: IR (CHCl<sub>3</sub>)  $\nu_{\max}$  3420, 2970, 2910, 1570, 1460, 1420, 1070, 936 cm<sup>-1</sup>. UV (MeOH)  $\lambda_{\max}$  211.5 nm ( $\epsilon$  20,100), EIMS  $m/z$  89 (C<sub>4</sub>H<sub>10</sub>ON+H)<sup>+</sup>, 103 (C<sub>5</sub>H<sub>12</sub>ON+H)<sup>+</sup>, 275, 277, 279 [1:2:1, (C<sub>8</sub>H<sub>7</sub>OBr<sub>2</sub>-2H)<sup>+</sup>], 290, 292, 294 [1:2:1, (C<sub>9</sub>H<sub>9</sub>OBr<sub>2</sub>-H)<sup>+</sup>], 334, 336, 338 (1:2:1, C<sub>11</sub>H<sub>14</sub>ONBr<sub>2</sub><sup>+</sup>). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  1.13 (3H, d,  $J$  = 6.3 Hz, H-1), 2.35 (1H, dd,  $J$  = 12.2, 10.7 Hz, H-3a), 2.43 (3H, s, H-13), 2.44 (1H, dd,  $J$  = 12.2, 2.9 Hz, H-3b), 2.59 (1H, ddd,  $J$  = 12.7, 11.2, 5.4 Hz, H-4a), 2.72 (1H, ddd, H-4b), 3.16 (1H, ddd,  $J$  = 12.7, 11.2, 5.4 Hz, H-5a), 3.25 (1H, ddd,  $J$  = 12.7, 11.2, 5.4 Hz, H-5b), 3.79 (1H, dqd,  $J$  = 10.3, 6.3, 2.9 Hz, H-2), 3.86 (3H, s, H-12), 6.65 (1H, d,  $J$  = 8.8 Hz, H-9), 7.45 (1H, d,  $J$  = 8.8 Hz, H-10).
- 8) 3: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  1.17 (3H, d,  $J$  = 6.3 Hz, H-1), 2.49 (1H, dd,  $J$  = 12.1, 9.2 Hz, H-3a), 2.82 (1H, dd,  $J$  = 12.2, 2.9 Hz, H-3b), 2.83 (1H, dt,  $J$  = 11.7, 8.0 Hz, H-4a), 2.88 (1H, dt,  $J$  = 11.7, 8.0 Hz, H-4b), 3.19 (2H, t, H-5), 3.80 (1H, dqd,  $J$  = 9.2, 6.2, 2.9 Hz, H-2), 3.86 (3H, s, H-12), 7.74 (1H, s, H-10).
- 9) 4: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  1.28 (3H, d,  $J$  = 6.4 Hz, H-1), 2.55 (1H, dd,  $J$  = 10.8, 7.8 Hz, H-3a), 3.24 (1H, dd,  $J$  = 10.8, 6.4 Hz, H-3b), 2.73 (2H, m, H-4), 3.19 (2H, m, H-5), 3.86 (3H, s, H-12), 4.15 (1H, dqd,  $J$  = 7.8, 6.4 Hz, H-2), 4.41 (1H, d,  $J$  = 4.9 Hz, H-13a), 4.44 (1H, d,  $J$  = 4.9 Hz, H-13b), 7.74 (1H, s, H-10).
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- 11) 5: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  1.28 (3H, d,  $J$  = 6.2 Hz, H-1), 1.40 (3H, s, H-15), 1.96 (1H, t,  $J$  = 11.0 Hz, H-3a), 2.23 (1H, d,  $J$  = 11.0 Hz, H-13a), 2.58 (2H, t,  $J$  = 8.0 Hz, H-4), 2.79 (1H, dd,  $J$  = 11.0, 2.5 Hz, H-13b), 2.87 (1H, dt,  $J$  = 11.0, 2.5 Hz, H-3b), 3.17 (2H, m, H-5), 3.87 (3H, s, H-12), 4.07 (1H, dqd, 11.0, 6.2, 2.5 Hz, H-2), 4.39 (1H, br s, OH-2), 7.74 (1H, s, H-10).
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